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thesineandcosinetotheappropriatenumberofdigits. $(31) 2^\circ \text{ rad}$
 $360 = 0.54 \text{ rad}$ $\sin(30.5) = 0.5075$ $\sin(31.5) = 0.5225$ $\sin(31) = 0.51$
 $\cos(30.5) = 0.86163$ $\cos(31.5) = 0.85264$ $\cos(31) = 0.86$
15. Some elementary chemistry textbooks give the value of R , the ideal gas constant, as $0.0821 \text{ l atm K}^{-1} \text{ mol}^{-1}$. a.

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v_{mean} M The mean translational kinetic energy is $\frac{3}{2}RT$ $3kT$
 $2 M^2$ The ratios of species 1 to species 2 at the same temperature are $\frac{v_{\text{mean},1}}{v_{\text{mean},2}}$ (i) $\frac{1}{2} \frac{m_2}{m_1}$
 $v_{\text{mean},H} = \frac{1}{2} \frac{m}{v^2}$ and $2 v_{\text{mean},Hg} = 200.6 \frac{4.003}{1} \frac{2}{2} \frac{m}{v_{\text{rms}}}$
 $E_k = \frac{1}{2} E_k = 7.079$ (ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

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