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$\text{Cu}(\text{NO}_3)_2 + \text{Zn} \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Cu}$ There is a sentence that you can write to describe what happens above: Copper(II) nitrate reacts with zinc metal and yields zinc nitrate and copper metal. Take a minute to notice the relationship between the sentence and the symbols in the reaction

~~Intro, to chemical Reactions - LTHS Answers~~

ChemQuest 35 Name: _____ Date: _____ Hour: _____ Information: Gas Pressure Figure 1: Two containers of gas molecules Container 1 Container 2 Gas molecules move randomly in their containers, colliding with the walls of their containers causing gas pressure. Pressure can be defined as the force pushing on an area. It can be described with the equation: where P is pressure (in kPa), F is ...

~~Information: Gas Pressure~~

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