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Specific Heat

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Calculations -

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~~Enthalpy Fusion,~~

~~Chemistry~~

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~~Calculating Latent~~

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~~How Much Thermal~~

~~Energy Is Required To~~

~~Heat Ice Into Steam-~~

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~~Answers~~

Worksheet-

Calculations

involving Specific

Heat 1. For $q = m c$

T : identify each

variables by name &

the units associated

with it. q = amount of

heat (J) m = mass

(grams) c = specific

heat (J/g ° C) $T =$

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Chemistry

change in
temperature ($^{\circ}\text{C}$) 2.

Specific Heat

Worksheet

Answers

Heat is not the same

as temperature, yet

they are related.

Explain how they

differ from each

other.

Worksheet

Calculations

involving Specific

Heat

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Heat worksheet

answers. Answers included on separate sheet. For $q = m c \Delta t$.

Also includes a spreadsheet to show how the calculations have been done. Use the data in the table to answer the following questions. Specific heat capacity handout answer key specific heat practice

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work. Explain how they differ from each other.

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~~Answers Worksheet List~~

Two page worksheet using Specific Heat Capacity. Questions start easy then become gradually harder. Answers

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included on separate sheet. Also includes a spreadsheet to show how the calculations have been done.

~~Specific Heat Capacity Worksheet (with answers) | Teaching ... Honors Chemistry Worksheet – Specific Heat. Recognize that when two systems at~~

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different temperatures meet, there will be a net transfer of heat (energy) from the system of greater heat intensity to the system of lower heat intensity. Summary – Heat flows from source to sink, in other words from hot to cold until thermal equilibrium is

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obtained. If you pick up a spoon sitting in some hot “ hot chocolate,” the spoon feels hot or warm because it is transferring heat to your body which ...

~~Honors Chemistry~~
~~Worksheet~~ ~~Specific~~
Heat

the boiling point if
ALL of the heat in

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Chemistry
Specific Heat
Worksheet
Answers

question (1) could be transferred to the water? The specific heat of water is $4.184 \text{ J/g} \cdot ^\circ\text{C}$. (3)

Magnesium metal has a specific heat of $1.04 \text{ J/g} \cdot ^\circ\text{C}$. A 70.0 g sample of this metal, at a temperature of $99.8 \text{ }^\circ\text{C}$, is added to a beaker containing 50.0 g of water at $30.0 \text{ }^\circ\text{C}$. The

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Specific Heat Chem
Worksheet 16 1

Answer Key.

Worksheet October
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Chemicals guide

covers a lot of

subjects, but here are
the quick and dirty

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details. The guide is broken down into four parts. The first one contains four sections and is called the Chemical Bonding Guide. In this part you learn how to recognize your bonding patterns and how to break them.

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~~Worksheet 16-1~~

~~Answer Key~~

Chemistry Specific
Heat Worksheet

Answers Honors

Chemistry Worksheet
– Specific Heat.

Recognize that when
two systems at
different

temperatures meet,
there will be a net
transfer of heat
(energy) from the

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system of greater heat intensity to the system of lower heat intensity. Summary – Heat flows from source to sink, in other words from hot ...

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Answers Chemistry:

States of Matter and

Specific Heat Review

Sheet ... Finding the

Specific Heat of a

Substance.

Chemistry-1 Lab:

Specific Heat Page 2

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Procedure: 1. If the hot plate you are sharing is not on, turn it on #8. The can should only have about 2 " – ...

Calculate

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Worksheet-
Specific Heat
Calculations

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involving Specific

Heat 1. For $q = m c$

T : identify each

variables by name &

the units associated

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heat (J) m = mass

(grams) c = specific

heat (J/g ° C) T =

change in

temperature (° C) 2.

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Specific Heat

Worksheet Answers

The specific heat capacity of aluminum is $0.90 \text{ J/g}\cdot\text{C}$. $Q = (10 \text{ g})(0.90)(55-22) = 297 \text{ J}$ 4.) Calculate the specific heat capacity for wood if $1500. \text{ g}$ of

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the wood absorbs
 6.75×10^4 Joules of
heat and its
temperature changes
from 32°C to 57°C
 $6.75 \times 10^4 \text{ J} = (1500 \text{ g})$
 $\text{C} (57-32) \text{ C} = 1.8$
 $\text{J/g}^\circ\text{C} 5.)$

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Calorimetry specific
heat and calculations~~

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The following is a list of specific heat capacities for a few metals. I have included printable pdf chemistry worksheets so you can practice problems and then check your answers.

~~Chemistry Worksheet
Heat And Calorimetry
Problems~~

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This covers specific heat capacity for P1 AQA. Lots of practice for using the SHC equation.

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Chemistry ...

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Calculations

involving Specific

Heat 1. For $q = m c$

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1. Identify each variable by name & the units associated with it. q = amount of heat (J) m = mass (grams) c = specific heat ($J/g\ ^\circ C$) T = change in temperature ($^\circ C$) 2.

~~Chemistry Specific Heat Worksheet~~
~~Answers~~

Specific Heat and

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Heat Capacity

Worksheet

DIRECTIONS: Use $q = (m)(C_p)(\Delta T)$ to solve the following

problems. Show all work and units. Ex:
How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C , if the specific heat of

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Chemistry
aluminum is 0.90

J/g ° C? 1.
Specific Heat

Worksheet

Specific Heat and
Heat Capacity

Worksheet

Purpose: Specific heat is a physical property. It measures how much energy (in Joules) is required to raise one gram of a substance by one degree Celsius. In this

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Worksheet, students will use the specific heat equation ($Q = mc \Delta T$) for a variety of different problems. Essential concepts: Specific heat, energy, Joules.

~~Heat Energy and
Enthalpy
Worksheets and
Lessons ...~~

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Answers

Answers: 1. 31.8oC 2.
52.0oC 3. 55.7oC 4.
41.1oC 5. 107.3 g 6.
168.6 g 7. tungsten 8.
A 97 g sample of gold
at 785oC is dropped
into 323 g of water,

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which has an initial temperature of 15°C . If gold has a specific heat of $0.129 \text{ J/g}\cdot^{\circ}\text{C}$, what is the final temperature of the mixture?

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947a705539

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