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ANSWER Answer

the following

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questions in the
space provided.

1. Pressure can
be represented
by the following
equation:

pressure

increase a. For
a constant area,
as the force
increases the
pressure will.

decrease b.

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area increases
the pressure

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Of Gases

following questions in the space provided.

1. Identify whether the descriptions below describe an ideal gas or a real gas. a. Gas particles move in straight lines until they collide with other particles

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or the walls of
their container.

Characteristics

Of Gases

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Molecular Theory
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Of Gases A.

Ideal Gas 1. An
imaginary gas
that perfectly
fits all the
assumptions of
the kinetic-
molecular theory

B. Five
Assumptions of
the Kinetic-
Molecular Theory

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1. Gases consist of large numbers of tiny particles that are far apart relative to their size 2.

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toys. Can the item be used with other toys. Will it help children reach program goals that are not being met sufficiently with current toys. Will help balance toys for all areas of development - ph

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Of Gases

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as the force
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Characteristics

of Gases. 1.

Chemistry S.

Martinez –

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2. KMT - based

on idea that

particles of

matter are

always in

motion. 2. 3. 1.

Gases consist of

large numbers of

tiny particles

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that are far
apart relative
to their size.2.
Of Gases

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