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being logical and wish to understand the real reason. As they are eager to learn what the ...

Balancing Chemical Equations Worksheet 1 Answer Key

Balancing Chemical Equations Worksheet 1 -
ANSWERS 1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2. $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3. $\text{N}_2\text{O}_4 \rightarrow 2\text{NO}_2$ 4. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ 5. $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ 6. $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$ 7. $2\text{Li} + \text{F}_2 \rightarrow 2\text{LiF}$ 8. $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$ 9. $2\text{NH}_3 \rightarrow 3\text{N}_2 + 2\text{H}_2$ 10. $2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$ 11. $2\text{NI}_3 \rightarrow \text{N}_2 + 3\text{I}_2$ 12. $2\text{HI} \rightarrow \text{H}_2 + \text{I}_2$ 13. $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$ 14.

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Balancing Chemical Equations Worksheet 1
Balancing Chemical Equations - Answer Key
Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 1 \text{ PbCl}_2 + 2 \text{ H}_2\text{O}$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$ 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$ 9) $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$ 10) $1 \text{ FeCl}_3 + 3 \text{ NaOH} \rightarrow 1 \text{ Fe(OH)}_3 + 3 \text{ NaCl}$

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Equations

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Answer Key

Word Equations - Answer Key 1) Zinc and lead (II) nitrate react to form zinc nitrate and lead. $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$ _____ 2) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas.

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If you also get perplexed in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

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Answer Key

Balancing Equations: Answers to Practice Problems 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers).

Balancing Equations: Practice Problems Solutions for the Balancing Equations Practice Worksheet. 1) $2 \text{NaNO}_3 + \text{PbO} \rightarrow \text{Pb}(\text{NO}_3)_2 + \text{Na}_2\text{O}$. 2) $6 \text{AgI} + \text{Fe}_2(\text{CO}_3)_3 \rightarrow 2 \text{FeI}_3 + 3 \text{Ag}_2\text{CO}_3$. 3) $\text{C}_2\text{H}_4\text{O}_2 + 2 \text{O}_2 \rightarrow 2 \text{CO}_2 + 2 \text{H}_2\text{O}$ - Answers. Balance these equations!

Note to students: Whenever balancing an equation, it is acceptable to leave spaces blank instead of writing "1" - in chemistry

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N2 + 3 NH3 1 L N20 Hgo Hg + 02 HCl c02 C

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Answer Key

$C_6H_{12}O_6 + I_2 + Zn + 2H_2SiO_4 + Na_2CO_3 + NaOH$
 $+ 1H_3PO_4 + H_4P_2O_7 + C + IU + HCl + ClO_4H_{16} +$
 $L) + CO_2 + Si_2H_3 + H_2SO_4 + i + 3Z + C + KOH + 3$
 $K_2SO_4 + 2C_7H_6O_2 + CO_2 + H_2O + 2H_2SO_4 + H_2S$
 $+ CO_2 + Al_2O_3 + FeO + Fe_2O_3 + H_2 + Fe + H_2O$
 $Na_2CO_3 + ZNaCl + ZKBr + O_2 + CO_2 + P_2O_5$

6. 9. 10. 11. 12. 13. 14. 16. 17. 18. 19. 20.
22. 23. 24 ...

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 Center Balancing Equations Worksheet -
 Solutions . 1) 1 $H_3PO_4 +$ 3 $KOH \rightarrow$ 1
 $K_3PO_4 +$ 3 H_2O . 2) 6 $K +$ 1 $B \dots$

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